

Facile synthesis of *N,N'*-dimethylated *N*-confused porphyrins

Ziwei Xiao and David Dolphin*

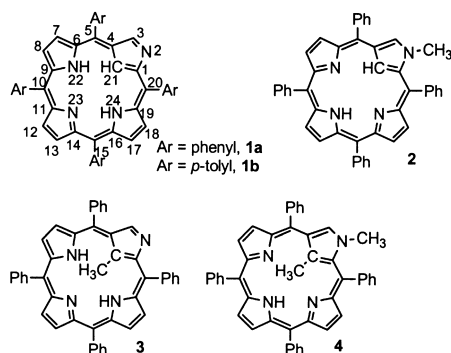
Department of Chemistry, University of British Columbia, 2036 Main Mall, Vancouver, BC, Canada V6T 1Z1

Received 6 June 2002; accepted 28 August 2002

Abstract—*N*-Confused tetraarylporphyrins react with CH₃I in the presence of Na₂CO₃ to give, in high yield, *N,N'*-dimethylated *N*-confused porphyrin salts, which are mixtures of structural isomers. The structures of the major isomers were determined by X-ray diffraction and NMR spectroscopic analyses. These *N,N'*-dimethylated *N*-confused tetraarylporphyrin salts generate singlet oxygen when irradiated at long wavelengths in the visible region, and therefore, are potential sensitizers for photodynamic therapy. © 2002 Elsevier Science Ltd. All rights reserved.

1. Introduction

In 1994, the groups of Furuta and Latos-Grazynski independently reported on the synthesis of *N*-confused porphyrins **1a**, **1b**.^{1,2} Since then the outer *N*-methylated *N*-confused porphyrin **2**, the *C*-methylated *N*-confused porphyrin **3**, the *C,N*-dimethylated *N*-confused porphyrin **4**, and their nickel(II) derivatives have also been prepared by Latos-Grazynski et al.^{3,4} We report here on the synthesis of *N,N'*-dimethylated *N*-confused porphyrins and their ability to generate singlet oxygen.



- 1a** *N*-confused tetraphenylporphyrin
1b *N*-confused tetra(*p*-tolyl)porphyrin
2 2-aza-2-methyl-5,10,15,20-tetraphenyl-21-carbaporphyrin
3 2-aza-21-methyl-5,10,15,20-tetraphenyl-21-carbaporphyrin
4 2-aza-2,21-dimethyl-5,10,15,20-tetraphenyl-21-carbaporphyrin

Keywords: *N,N'*-dimethylated *N*-confused porphyrin; *N*-confused porphyrin; methylation; singlet oxygen.

* Corresponding author. Tel.: +1-604-822-4571; fax: +1-604-822-9678; e-mail: ddolphin@q1tinc.com

2. Results and discussion

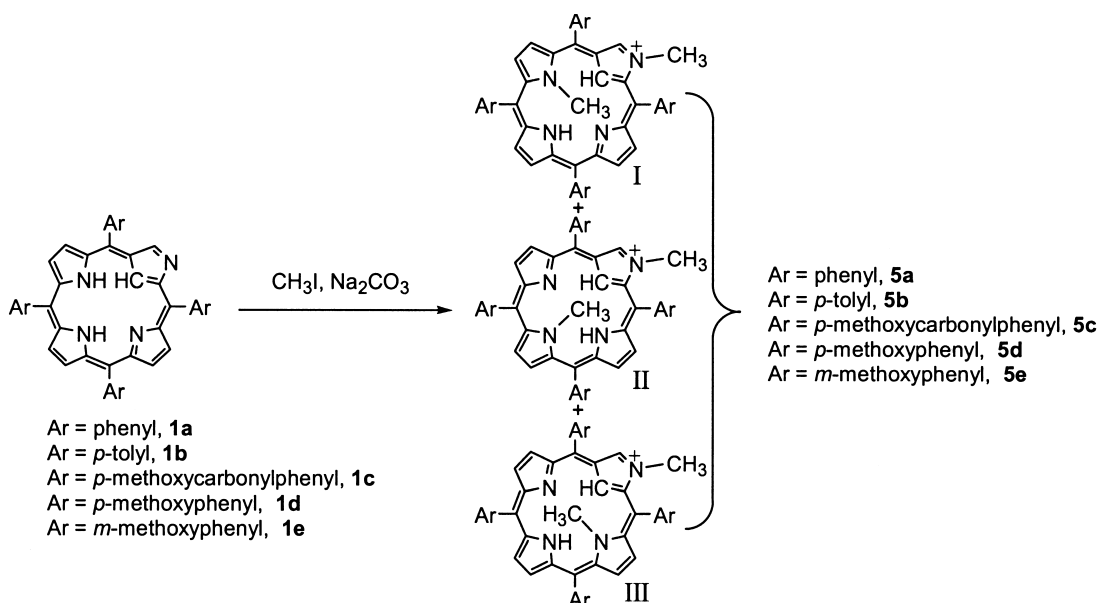
2.1. Formation of *N,N'*-dimethylated *N*-confused porphyrins

The methylation of **1a** with CH₃I in CH₂Cl₂ yields compound **2**.³ It is assumed that protonation of compound **2** by HI generated from the methylation reaction prevents subsequent inner *N*-methylation. Therefore, addition of a base, such as Na₂CO₃, might neutralize the protonated intermediate, making the inner nitrogens more nucleophilic towards methylation and resulting in *N,N'*-dimethylation. In fact, near-quantitative dimethylation of *N*-confused porphyrins, as shown by TLC, was carried out in CH₂Cl₂ solution using CH₃I and Na₂CO₃. The *N,N'*-dimethylated *N*-confused porphyrin salts have a broad absorption around 790 nm (~20000 M⁻¹ cm⁻¹), and therefore, will be of interest for their use as photosensitizers in photodynamic therapy (PDT).⁵

However, each of the *N,N'*-dimethylated *N*-confused porphyrin salts **5(a–e)** may have three structural isomers (I–III, Scheme 1), since there are three possible positions, N-22, N-23, and N-24, for inner *N*-methylation. Separation of the isomers using various chromatographic techniques was not successful in our hands. Only the major isomer of compounds **5(a–d)** could be isolated by recrystallization.

2.2. Crystal structure of compound **5a-III**

Compound **5a-III** (Fig. 1) is the major isomer of *N,N'*-dimethylated 2-aza-5,10,15,20-tetraphenyl-21-carbaporphyrin-HI (**5a**). When the I⁻ was exchanged for CF₃SO₃⁻ in **5a-III**, the structure was determined by X-ray crystallography, which confirms that the porphyrin is *N,N'*-dimethylated.



Scheme 1. The synthesis of *N,N'*-dimethylated 2-aza-5,10,15,20-tetraphenyl-21-carbaporphyrin- I^- (**5a**), *N,N'*-dimethylated 2-aza-5,10,15,20-tetra(*p*-tolyl)-21-carbaporphyrin- I^- (**5b**), *N,N'*-dimethylated 2-aza-5,10,15,20-tetrakis(*p*-methoxycarbonylphenyl)-21-carbaporphyrin- I^- (**5c**), *N,N'*-dimethylated 2-aza-5,10,15,20-tetrakis(*p*-methoxyphenyl)-21-carbaporphyrin- I^- (**5d**) and *N,N'*-dimethylated 2-aza-5,10,15,20-tetrakis(*m*-methoxyphenyl)-21-carbaporphyrin- I^- (**5e**).

2.3. Structural determination by NMR spectroscopy

The structure of **5d-III**, the major isomer of **5d**, was determined by NMR spectroscopic analyses (1H , ^{13}C , selective NOE, HMQC and HMBC (Fig. 2)). The 1H 3.81 ppm peak was assigned to H-25 based on the observed cross peak with the ^{13}C 39.5 ppm peak using HMQC. Selective NOE experiments showed correlations of H-25 (3.81 ppm) with H-43a (8.40 ppm) and H-43b (8.04 ppm), correlations of H-43a with H-43b and H-18 (7.56 ppm) and correlation of H-18 with H-17 (7.16 ppm). The 2-*N*-methyl group hinders rotation of the adjacent *p*-methoxyphenyl group. The rotation is slow and H-43a and H-43b can be distinguished by NMR spectroscopy. However, there is still some rotation, which results in H-43a and H-43b being interchangeable and their peaks broadened. The negative

peak at 8.04 ppm, upon irradiation at 8.40 ppm, showed that the two hydrogens are interchangeable. The -1.47 ppm peak was assigned to the inner methyl group, as suggested by the chemical shift and integration. The observed cross peaks of ^{13}C 150.8 ppm with H-17 (7.16 ppm), H-18 (7.56 ppm) and H-26 (-1.47 ppm) in an HMBC experiment clearly showed that the inner methyl group is connected to the pyrrole unit containing H-17 and H-18.

The structures of the major isomers of compounds **5b** and **5c** were determined in a similar way by NMR spectroscopic analysis. In each case, the inner methylation occurs on N-24, the same as for **5a-III** and **5d-III**.

Because the 2-*N*-methyl group hinders rotations of the adjacent *m*-methoxyphenyl group, compounds **5e** has

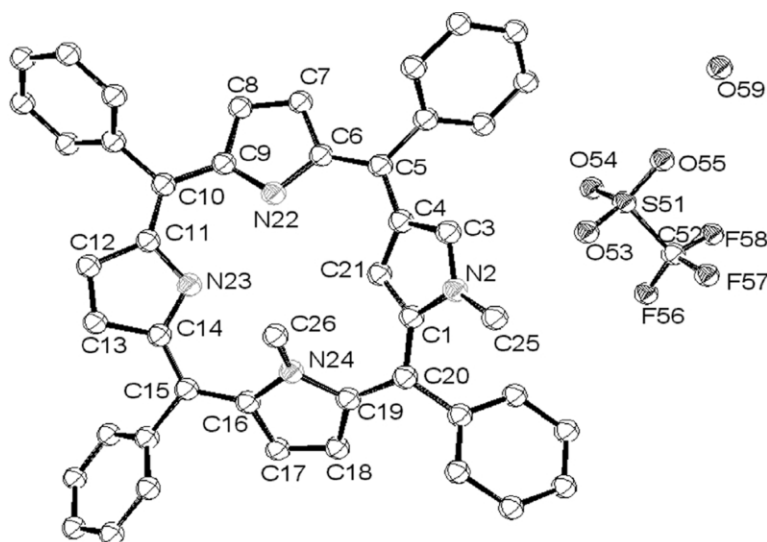


Figure 1. ORTEP drawing of compound (**5a-III**) showing atomic labeling and thermal ellipsoids at the 50% probability level.

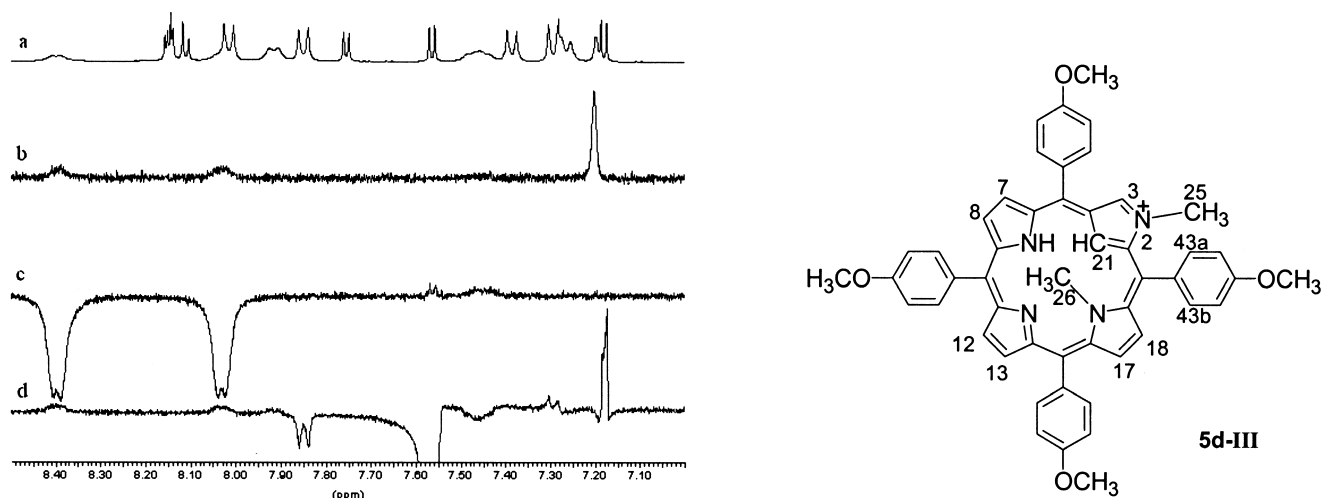


Figure 2. ^1H NMR spectra of **5d-III** in CD_2Cl_2 : (a) no irradiation; (b) upon irradiation at 3.81 ppm; (c) upon irradiation at 8.40 ppm; (d) upon irradiation at 7.56 ppm.

atropisomers in addition to structural isomers, making it a more complicated mixture than compound **5(a–d)**. Purification by recrystallization was not successful in this case.

2.4. Generation of singlet oxygen

1,3-Diphenylisobenzofuran (DPBF)⁶ was used to determine the ability of these N,N' -dimethylated N -confused porphyrin salts to generate singlet oxygen. DPBF reacts quickly with singlet oxygen and its absorption decay around 418 nm can be readily monitored. The reaction products of DPBF have no absorption in the visible region and do not quench singlet oxygen. A solution containing DPBF ($\sim 18 \mu\text{M}$) and N,N' -dimethylated N -confused porphyrin salts (e.g. **5b**, $\sim 13 \mu\text{M}$) was irradiated with a halogen lamp using a filter (700 nm) and monitored by UV–vis spectroscopy at 418 nm. Substantial decay of the UV–vis signal at

418 nm was observed confirming that N,N' -dimethylated N -confused porphyrin salts generate singlet oxygen (Fig. 3).

3. Conclusions

A series of N,N' -dimethylated N -confused tetraarylporphyrin salts were synthesized through methylation of N -confused tetraarylporphyrins using CH_3I in the presence of Na_2CO_3 . The inner methylation of the major isomers occurs on N-24, as determined by X-ray diffraction and NMR spectroscopic analyses. These compounds have a broad absorption around 790 nm ($\sim 20,000 \text{ M}^{-1} \text{ cm}^{-1}$). It has also been shown that N,N' -dimethylated N -confused tetraarylporphyrin salts can generate singlet oxygen when irradiated, suggesting that they are potential sensitizers for PDT.

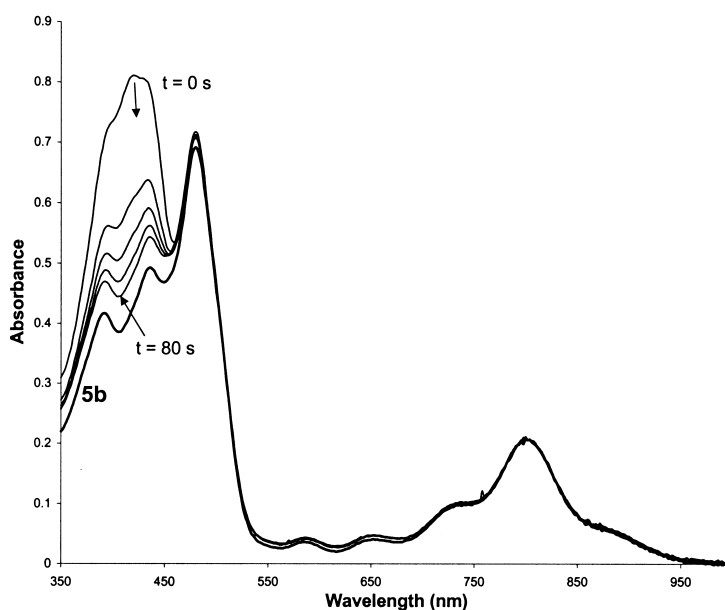


Figure 3. UV–vis spectra of a solution of DPBF and compound **5b** before and after irradiation (irradiation intervals at 20 s). The UV–vis spectrum of pure compound **5b** is shown in a thicker line.

4. Experimental

4.1. General information

Pyrrole (Acros) was distilled from CaH₂ before use. The silica gel was 230–400 mesh (Silicycle). Activity III basic alumina was obtained by adding 6% water to activity I Brockman basic alumina, 60–325 mesh (Fisher). All other materials and solvents were used as received. The NMR spectra were recorded on a Bruker WH-400, a Bruker AV-400 or a Bruker AMX-500 in the solvents indicated and were referenced to residual solvent peaks. Elemental analyses were performed on a Carlo Erba Elemental Analyzer 1108. The UV–vis spectra were measured on a Cary 50. Mass spectra were determined on a KRATOS Concept IIIHQ hybrid mass spectrometer. Irradiations were carried with a 250 W Osram HLX 64655 arc lamp in an Oriel lamp housing (model 66184). The light output passed through a filter: P70-700-S-Corion.

4.2. *N*-Confused porphyrins

N-Confused porphyrins were synthesized using a modification of Lindsey's procedure.⁷ To a solution of pyrrole (0.52 mL, 7.5 mmol) and arylaldehyde (7.5 mmol) in CH₂Cl₂ (750 mL) was added methanesulfonic acid (MSA) (0.34 mL, 5.2 mmol). The mixture was stirred for 30 min after which DDQ (1.50 g, 6.6 mmol) was added. After 1 min, triethylamine (1.5 mL) was added. The crude reaction mixture was chromatographed with silica gel (14×4.4 cm) under vacuum and eluted with CH₂Cl₂. CH₂Cl₂/1.2% methanol eluted the product with impurities. The fractions were collected and concentrated under vacuum and then absorbed onto 7.5 g of activity III basic alumina. The absorbed sample was added to the top of a column with 150 g activity III basic alumina in 2:1 hexanes/CH₂Cl₂. The polarity of the eluant was increased from 2:1 to 1:1 to 1:2 hexanes/CH₂Cl₂, the *N*-confused porphyrins were eluted with 1:2 hexanes/CH₂Cl₂ (in the case of compounds **1c** and **1d**, the polarity of the eluant was increased from CH₂Cl₂ to CH₂Cl₂/0.2% methanol to elute the product). The solvent was removed under vacuum and the residue was triturated with CH₂Cl₂/hexanes to yield the product. Compound **1a**, yield: 373 mg (32%); compound **1b**, yield: 274 mg (22%); compound **1c**, yield: 259 mg (16%); compound **1d**, yield: 208 mg (15%); compound **1e**, yield: 316 mg (23%).

4.2.1. 2-Aza-5,10,15,20-tetrakis(*p*-methoxycarbonylphenyl)-21-carbaporphyrin (1c**).** *R_f* (silica–CH₂Cl₂/5% CH₃OH/2% Et₃N) 0.46; ¹H NMR (400 MHz, CD₂Cl₂) δ = –5.24 (s, 1H), –2.56 (br s, 2H), 4.08 (d, 12H), 8.18–8.58 (m, 19H), 8.60 (d, *J* = 4.7 Hz, 1H), 8.70 (s, 1H), 8.84 (d, *J* = 4.7 Hz, 1H), 8.94 (d, *J* = 4.7 Hz, 1H); UV–vis (CH₂Cl₂) λ_{max}/nm (log ε) 444 (5.31), 544 (4.07), 586 (4.22), 728 (4.15); MS (LSIMS) 847 (MH⁺, 100%); HRMS (LSIMS) *m/e* calcd for C₅₂H₃₀N₄O₈: 847.27679, found 847.27667 (MH⁺). Anal. calcd for C₅₂H₃₈N₄O₈: C, 73.75; H, 4.52; N, 6.62. Found: C, 73.95; H, 4.49; N, 6.72.

4.2.2. 2-Aza-5,10,15,20-tetrakis(*p*-methoxyphenyl)-21-carbaporphyrin (1d**).** *R_f* (silica–CH₂Cl₂/5% CH₃OH/2% Et₃N) 0.38; ¹H NMR (400 MHz, CD₂Cl₂) δ = –4.92 (s, 1H),

–2.31 (br s, 2H), 4.04 (s, 3H), 4.07 (s, 6H), 4.09 (s, 3H), 7.25–7.49 (m, 8H), 8.00–8.13 (m, 4H), 8.25 (d, *J* = 8.6 Hz, 4H), 8.55 (d, *J* = 5.2 Hz, 3H), 8.61 (d, *J* = 4.7 Hz, 1H), 8.64 (s, 1H), 8.89 (d, *J* = 4.4 Hz, 1H), 8.97 (d, *J* = 4.7 Hz, 1H); UV–vis (CH₂Cl₂) λ_{max}/nm (log ε) 442 (5.38), 514 (sh), 548 (sh), 592 (4.32), 736 (4.24); MS (LSIMS) 735 (MH⁺, 100%); HRMS (LSIMS) *m/e* calcd for C₄₈H₃₉N₄O₄: 735.29713, found 735.29721 (MH⁺). Anal. calcd for C₄₈H₃₈N₄O₄: C, 78.45; H, 5.21; N, 7.62. Found: C, 78.11; H, 5.10; N, 7.71.

4.2.3. 2-Aza-5,10,15,20-tetrakis(*m*-methoxyphenyl)-21-carbaporphyrin (1e**).** *R_f* (silica–CH₂Cl₂/5% CH₃OH/2% Et₃N) 0.50; ¹H NMR (400 MHz, CD₂Cl₂) δ = –5.12 (s, 1H), –2.45 (br s, 2H), 3.98 (s, 6H), 4.02 (s, 3H), 4.06 (s, 3H), 7.22–7.45 (m, 4H), 7.58–7.82 (m, 8H), 7.82–8.00 (m, 4H), 8.55–8.66 (m, 3H), 8.68 (d, *J* = 4.7 Hz, 1H), 8.78 (s, 1H), 8.96 (d, *J* = 4.7 Hz, 1H), 9.05 (d, *J* = 5.2 Hz, 1H); UV–vis (CH₂Cl₂) λ_{max}/nm (log ε) 440 (5.34), 540 (4.06), 582 (4.13), 726 (4.11); MS (LSIMS) 735 (MH⁺, 100%); HRMS (LSIMS) *m/e* calcd for C₄₈H₃₉N₄O₄: 735.29713, found 735.29827 (MH⁺). Anal. calcd for C₄₈H₃₈N₄O₄: C, 78.45; H, 5.21; N, 7.62. Found: C, 78.44; H, 5.18; N, 7.71.

4.3. General method for synthesis of the *N,N'*-dimethylated *N*-confused porphyrins

N-Confused porphyrin (100 mg) was dissolved in a minimal amount of CH₂Cl₂ (about 10 mL). To this solution, CH₃I (8 mL) and Na₂CO₃ (250 mg) were added. The mixture was stirred for 2 days in the absence of light, then filtered through Celite. The filtrate was evaporated to dryness under vacuum and the residue was triturated with CH₂Cl₂/hexanes to yield the products.

4.3.1. *N,N'*-Dimethylated 2-aza-5,10,15,20-tetraphenyl-21-carbaporphyrin-I[–] (5a**).** Yield: 110 mg (88%). *R_f* (silica–CH₂Cl₂/5% CH₃OH/2% Et₃N) 0.36; ¹H NMR (400 MHz, CD₂Cl₂) δ = –1.62 (d, *J* = 1.4 Hz, 0.65H), –1.57 (d, *J* = 1.5 Hz, 0.35H), –1.48 (s, 3×0.65H), –1.39 (s, 3×0.35H), 3.62 (s, 3×0.35H), 3.82 (s, 3×0.65H), 7.18–8.54 (m, 27H); UV–vis (CH₂Cl₂) λ_{max}/nm (log ε) 382 (sh), 476 (4.85), 582 (3.79), 650 (3.73), 724 (sh), 788 (4.39); MS (LSIMS) 643 (M⁺, 100%); HRMS (LSIMS) *m/e* calcd for C₄₆H₃₅N₄: 643.28617, found 643.28623 (M⁺). Anal. calcd for C₄₆H₃₅N₄·1.5H₂O: C, 69.26; H, 4.80; N, 7.02; I, 15.91. Found: C, 68.96; H, 4.76; N, 6.86; I, 15.82.

4.3.2. 2-Aza-2,24-dimethyl-5,10,15,20-tetraphenyl-21-carbaporphyrin-CF₃SO₃[–] (5a-III**).** Compound **5a** (86 mg) was dissolved in 20 mL of CH₂Cl₂. Silver triflate (1.3 g) was added and the solution was stirred for 2 h. The mixture was chromatographed through a silica gel column (10×2.5 cm) and eluted with CH₂Cl₂ under vacuum. CH₂Cl₂/1% CH₃OH eluted the porphyrin triflate salts. The compound was recrystallized three times with CH₂Cl₂/hexanes giving **5a-III** (29 mg). Crystals of **5a-III** were obtained by solvent diffusion of hexanes into CH₂Cl₂. *R_f* (silica–CH₂Cl₂/5% CH₃OH/2% Et₃N) 0.36; ¹H NMR (400 MHz, CD₂Cl₂) δ = –1.60 (s, 1H), –1.49 (s, 3H), 3.81 (s, 3H), 7.24 (d, *J* = 5.0 Hz, 1H), 7.43 (s, 1H), 7.62 (d, *J* = 5.0 Hz, 1H), 7.70–8.13 (m, 20H), 8.16 (d, *J* = 5.1 Hz, 1H), 8.20 (d, *J* = 5.1 Hz, 1H), 8.33 (d, *J* = 4.8 Hz, 1H), 8.47 (d,

$J=6.4$ Hz, 1H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 385 (sh), 475 (4.85), 580 (3.74), 650 (3.67), 724 (sh), 790 (4.41); MS (LSIMS) 643 (M^+ , 100%). Anal. calcd for $\text{C}_{46}\text{H}_{35}\text{N}_4\cdot\text{CF}_3\text{SO}_3\cdot 1.5\text{H}_2\text{O}$: C, 68.85; H, 4.67; N, 6.83; S, 3.91. Found: C, 68.99; H, 4.61; N, 6.78; S, 4.00.

Crystal data for 5a-III. $\text{C}_{47}\text{H}_{35}\text{N}_4\text{SO}_3\text{F}_3\cdot\text{H}_2\text{O}$, $M=810.89$, triclinic, $a=11.922(2)$ Å, $b=13.505(2)$ Å, $c=15.090(3)$ Å, $\alpha=110.043(6)^\circ$, $\beta=94.106(6)^\circ$, $\gamma=113.859(7)^\circ$, $V=2023.9(6)$ Å³, $T=198.2$ K, space group $P\bar{1}$ (#2), $Z=2$, $\mu(\text{Mo K}\alpha)=1.44$ cm⁻¹, 11514 reflections measured, 5815 unique ($R_{\text{int}}=0.056$) which were used in all calculations. The final $R_w(F^2)$ was 0.176 (all data).

4.3.3. *N,N'*-Dimethylated 2-aza-5,10,15,20-tetrakis(*p*-tolyl)-21-carbaporphyrin-I⁻ (5b). Yield: 105 mg (85%). R_f (silica– $\text{CH}_2\text{Cl}_2/5\%$ $\text{CH}_3\text{OH}/2\%$ Et_3N) 0.36; ¹H NMR (400 MHz, CD_2Cl_2) $\delta=-1.53$ (d, $J=1.7$ Hz, 0.65H), -1.48 (s, $3\times 0.65\text{H}$), -1.45 (d, $J=1.7$ Hz, 0.35H), -1.34 (s, $3\times 0.35\text{H}$), 2.65 (m, 12H), 3.61 (s, $3\times 0.35\text{H}$), 3.81 (s, $3\times 0.65\text{H}$), 7.1–8.5 (m, 23H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 392 (sh), 478 (4.85), 588 (3.64), 646 (3.69), 798 (4.22); MS (LSIMS) 699 (M^+ , 100%); HRMS (LSIMS) *m/e* calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4$: 699.34877, found 699.34859 (M^+). Anal. calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4\cdot\text{I}\cdot 0.5\text{H}_2\text{O}$: C, 71.85; H, 5.31; N, 6.70. Found: C, 71.85; H, 5.41; N, 6.70.

4.3.4. 2-Aza-2,24-dimethyl-5,10,15,20-tetrakis(*p*-tolyl)-21-carbaporphyrin-I⁻ (5b-III). Compounds **5b** (109 mg) was recrystallized three times with $\text{CH}_2\text{Cl}_2/\text{hexanes}$ to give the major isomer (34 mg). R_f (silica– $\text{CH}_2\text{Cl}_2/5\%$ $\text{CH}_3\text{OH}/2\%$ Et_3N) 0.36; ¹H NMR (400 MHz, CD_2Cl_2) $\delta=-1.55$ (d, $J=1.7$ Hz, 1H), -1.49 (s, 3H), 2.63 (s, 6H), 2.66 (s, 6H), 3.82 (s, 3H), 7.22 (d, $J=5.2$ Hz, 1H), 7.33 (d, $J=1.3$ Hz, 1H), 7.51–8.05 (m, 17H), 8.15 (d, $J=5.2$ Hz, 1H), 8.18 (d, $J=5.2$ Hz, 1H), 8.29 (d, $J=4.7$ Hz, 1H), 8.34 (br s, 1H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 390 (sh), 435 (sh), 480 (4.78), 585 (3.54), 650 (3.56), 800 (4.26); MS (LSIMS) 699 (M^+ , 100%). Anal. calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4\cdot\text{I}\cdot\text{H}_2\text{O}$: C, 71.08; H, 5.37; N, 6.63; I, 15.02. Found: C, 71.05; H, 5.31; N, 6.58; I, 14.95.

4.3.5. *N,N'*-Dimethylated 2-aza-5,10,15,20-tetrakis(*p*-methoxycarbonylphenyl)-21-carbaporphyrin-I⁻ (5c). Yield: 103 mg (87%). R_f (silica– $\text{CH}_2\text{Cl}_2/5\%$ $\text{CH}_3\text{OH}/2\%$ Et_3N) 0.36; ¹H NMR (400 MHz, CD_2Cl_2) $\delta=-1.79$ (m, 1H), -1.49 (m, 3H), 3.65 (s, $3\times 0.27\text{H}$), 3.87 (s, $3\times 0.73\text{H}$), 4.05 (m, 12H), 7.20–8.73 (m, 23H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 386 (sh), 478 (4.84), 576 (3.86), 652 (3.81), 726 (sh), 792 (4.24); MS (LSIMS) 875 (M^+ , 100%); HRMS (LSIMS) *m/e* calcd for $\text{C}_{54}\text{H}_{43}\text{N}_4\text{O}_8$: 875.30809, found 875.30815 (M^+). Anal. calcd for $\text{C}_{54}\text{H}_{43}\text{N}_4\text{O}_8\cdot\text{I}\cdot 2.5\text{H}_2\text{O}$: C, 61.89; H, 4.62; N, 5.35. Found: C, 61.99; H, 4.42; N, 5.20.

4.3.6. 2-Aza-2,24-dimethyl-5,10,15,20-tetrakis(*p*-methoxycarbonylphenyl)-21-carbaporphyrin-I⁻ (5c-III). Compounds **5c** (78.5 mg) was recrystallized three times with $\text{CH}_2\text{Cl}_2/\text{hexanes}$ to give the major isomer (29 mg). R_f (silica– $\text{CH}_2\text{Cl}_2/5\%$ $\text{CH}_3\text{OH}/2\%$ Et_3N) 0.36; ¹H NMR (400 MHz, CDCl_3) $\delta=-1.45$ (d, 4H), 3.91 (s, 3H), 4.06 (s, 12H), 7.19 (s, 1H), 7.52 (s, 2H), 7.82–8.85 (m, 20H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 385 (sh), 480 (4.83), 585

(3.67), 655 (3.16), 725 (sh), 790 (4.26); MS (LSIMS) 875 (M^+ , 100%). Anal. calcd for $\text{C}_{54}\text{H}_{43}\text{N}_4\text{O}_8\cdot\text{I}\cdot\text{H}_2\text{O}$: C, 63.53; H, 4.44; N, 5.49; I, 12.43. Found: C, 63.26; H, 4.55; N, 5.48; I, 12.20.

4.3.7. *N,N'*-Dimethylated 2-aza-5,10,15,20-tetrakis(*p*-methoxyphenyl)-21-carbaporphyrin-I⁻ (5d). Yield: 112 mg (92%). R_f (silica– $\text{CH}_2\text{Cl}_2/5\%$ $\text{CH}_3\text{OH}/2\%$ Et_3N) 0.36; ¹H NMR (400 MHz, CD_2Cl_2) $\delta=-1.43$ (s, $3\times 0.7\text{H}$), -1.37 (d, $J=1.5$ Hz, $1\times 0.7\text{H}$), -1.22 (d, $4\times 0.3\text{H}$), 3.60 (s, $3\times 0.3\text{H}$), 3.82 (s, $3\times 0.7\text{H}$), 4.00–4.14 (m, 12H), 7.07–8.50 (m, 23H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 436 (4.82), 486 (4.97), 592 (3.62), 662 (3.82), 818 (4.40); MS (LSIMS) 763 (M^+ , 100%). Anal. calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4\text{O}_4\cdot\text{I}\cdot 1.5\text{H}_2\text{O}$: C, 65.43; H, 5.05; N, 6.10. Found: C, 65.67; H, 4.88; N, 5.98.

4.3.8. 2-Aza-2,24-dimethyl-5,10,15,20-tetrakis(*p*-methoxyphenyl)-21-carbaporphyrin-I⁻ (5d-III). Compounds **5d** (89 mg) was recrystallized three times with $\text{CH}_2\text{Cl}_2/\text{hexanes}$ to give the major isomer **5d-III** (37 mg). R_f (silica– $\text{CH}_2\text{Cl}_2/5\%$ $\text{CH}_3\text{OH}/2\%$ Et_3N) 0.36; ¹H NMR (500 MHz, CDCl_3) $\delta=-1.47$ (s, 3H), -1.44 (s, 1H), 3.81 (s, 3H), 4.01–4.13 (m, 12H), 7.16 (d, $J=4.9$ Hz, 1H), 7.19 (s, 1H), 7.28 (m, 4H), 7.38 (d, $J=8.6$ Hz, 2H), 7.46 (m, 2H), 7.56 (d, $J=4.9$ Hz, 1H), 7.75 (d, $J=4.7$ Hz, 1H), 7.84 (d, $J=8.7$ Hz, 2H), 7.91 (d, $J=7.0$ Hz, 2H), 8.04 (m, 3H), 8.11 (d, $J=5.1$ Hz, 1H), 8.14 (d, $J=4.6$ Hz, 1H), 8.15 (d, $J=5.1$ Hz, 1H), 8.40 (d, $J=6.4$ Hz, 1H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 436 (4.78), 486 (4.91), 590 (3.56), 658 (3.74), 824 (4.37); MS (LSIMS) 763 (M^+ , 100%); HRMS (LSIMS) *m/e* calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4\text{O}_4$: 763.32843, found 763.32857 (M^+). Anal. calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4\text{O}_4\cdot\text{I}$: C, 67.40; H, 4.87; N, 6.29; I, 14.25. Found: C, 67.11; H, 4.97; N, 6.16; I, 14.09.

4.3.9. *N,N'*-Dimethylated 2-aza-5,10,15,20-tetrakis(*m*-methoxyphenyl)-21-carbaporphyrin-I⁻ (5e). Yield: 99 mg (82%). R_f (silica– $\text{CH}_2\text{Cl}_2/5\%$ $\text{CH}_3\text{OH}/2\%$ Et_3N) 0.36; ¹H NMR (400 MHz, CD_2Cl_2) $\delta=-1.61$ (m, 1H), -1.48 (d, $3\times 0.7\text{H}$), -1.39 (s, $3\times 0.3\text{H}$), 3.68 (s, $3\times 0.3\text{H}$), 3.86 (s, $3\times 0.7\text{H}$), 3.95–4.17 (m, 12H), 7.21–8.42 (m, 23H); UV–vis (CH_2Cl_2) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 378 (sh), 478 (4.95), 584 (3.81), 654 (3.74), 724 (sh), 788 (4.38); MS (LSIMS) 763 (M^+ , 100%); HRMS (LSIMS) *m/e* calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4\text{O}_4$: 763.32843, found 763.32854 (M^+). Anal. calcd for $\text{C}_{50}\text{H}_{43}\text{N}_4\text{O}_4\cdot\text{I}\cdot 0.5\text{H}_2\text{O}$: C, 66.74; H, 4.93; N, 6.23; I, 14.10. Found: C, 66.72; H, 4.93; N, 6.30; I, 13.97.

4.4. Singlet oxygen tests

A solution containing DPBF and an *N,N'*-dimethylated *N*-confused porphyrin salt (one of compounds **5(a–e)** or the major isomer of **5(a–d)**) ($\text{OD}=0.8$ – 1.0 at 418 nm, $\text{OD}=0.1$ – 0.2 at irradiation wavelength) was prepared and the UV–vis spectra were measured. The solution was then irradiated with a halogen lamp using a filter (700 nm) for four 20 s intervals and UV–vis spectra were taken after each interval. Substantial decay of the signal around 418 nm was observed in each case. No change in UV–vis spectra was observed after a sample containing DPBF and an *N,N'*-dimethylated *N*-confused porphyrin salts was left in the dark for 10 min, and there is also no change in UV–vis

spectra after irradiating a solution containing only DPBF or an *N,N'*-dimethylated *N*-confused porphyrin salts for 1 min.

4.5. Supporting information available

Crystallographic data (excluding structure factors) for the structures in this paper have been deposited with the Cambridge Crystallographic Data Center as supplementary publication number CCDC 169756.

Acknowledgements

This work was supported by the Natural Sciences and Engineering Research Council (NSERC) of Canada. We thank Dr Brian O. Patrick for X-ray diffraction analysis.

References

1. Furuta, H.; Asano, T.; Ogawa, T. *J. Am. Chem. Soc.* **1994**, *116*, 767–768.
2. Chmielewski, P. J.; Latos-Grazynski, L.; Rachlewicz, K.; Glowiac, T. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 779–781.
3. Chmielewski, P. J.; Latos-Grazynski, L. *J. Chem. Soc., Perkin Trans. 2* **1995**, 503–509.
4. Chmielewski, P. J.; Latos-Grazynski, L.; Glowiac, T. *J. Am. Chem. Soc.* **1996**, *118*, 5690–5701.
5. (a) MacDonald, I. J.; Dougherty, T. J. *J. Porphyr. Phthalocya.* **2001**, *5*, 145–158. (b) Sternberg, E.; Dolphin, D. *Curr. Med. Chem.* **1996**, *3*, 293–324.
6. Spiller, W.; Kliesch, H.; Wohrle, D.; Hackbarth, S.; Roder, B.; Schnurpfeil, G. *J. Porphyr. Phthalocya.* **1998**, *2*, 145–158.
7. Geier, III., G. R.; Haynes, D. M.; Lindsey, J. S. *Org. Lett.* **1999**, *1*, 1455–1458.